

Nachhaltige Produktion heutiger und zukünftiger Batteriegenerationen mit zirkulärer Wertschöpfung von Produktionsabfall und EoL-Batterien

Prof. Dr.-Ing Arno Kwade, Institut für Partikeltechnik & Battery LabFactory Braunschweig

Content



- **2** Ways to reduce CO_2 -Impact in Production
- **3** Closed-loop Circulation of Production Scrap & EoL Batteries
 - Perspective Circular Production of Solid State Batteries

5 Conclusion and Outlook





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- Diversification of cell technologies, sizes and designs
- Different generations of high performance batteries will coexist
- Lower cost battery cells with better ecological footprint but lower performance will also be developed



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Doose, S. et al. *Metals* 2021, 11, 291. <u>https://doi.org/10.3390/met11020291</u>

For EVs the environmental impacts shift to the production Importance to avoid problem shifting Institute for Particle Technology



Cradle-to-gate CO₂ emissions of NMC batteries

60% of productionrelated CO₂-emissions are embodied in the materials of an average EV battery



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Data from Hawkins et al. (2013), Comparative Environmental Life Cycle Assessment of Conventional and Electric Vehicles, Journal of Industrial Ecology, 17(1):53-64 and Ciez R E, Whitacre J F (2019) Examining different recycling processes for lithium-ion batteries. Nature Sustainability, 2(2):148-156.





Demand on raw materials for electric vehicles

Metric tons x 10³ 8,000 Mangan 7,000 Lithium 6,000 Cobalt 5,000 4,000 Nickel 3,000 Graphite 2,000 Aluminium 1,000 Cupper 0 2021 2024 2025 2026 2027 2028 2029 2030 2018 2019 2020 2022 2023

Source: Electric Vehicle Outlook 2018, Bloomberg New Energy Finance. Note: Copper includes copper current collectors and pack wiring. Aluminium includes aluminium current collectors, cell and pack materials and aluminium in cathode active materials.



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- CO₂-impact of materials has to be minimized
- Social aspects of raw materials have to be taken into account



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Measures to reach a sustainable battery cell



Potentials in active and passive material production



DOI: https://doi.org/10.1039/C4EE03029J



- Active / inactive material synthesis offers potential to reduce the CO2 footprint
- Cathode material in particular shows high energy-saving potential
- Depending on the specific capacity of the active material, the energy consumption and CO2-footprint per kg of battery can be improved
- Synthesis and processing to battery materials must be considered further (e.g. cathode active materials for , LFP aqueous processing – LFP, coated NMC)







Potentials of Sustainable Cell Design



- Water based binder, if possible based on biopolymers
- Cathode materials without or minimum amount of cobalt and nickel
- Thick electrodes to reduce relative content of Aluminium and Cupper
- Change of cell chemistry, e.g. towards Li-Sulfur or Sodium-Ion-Battery





Potentials of sustainable battery cell manufacturing

Process chain of lithium-ion battery cell manufacturing



battery production technologies. Nature Energy, vol. 3, pages 290-300 (2018)

Established electrode production Different wet process routes





Sustainable electrode production

Continuous extrusion

Batch



Processing time: > 2 h Maximum solids content: 75 wt.-%

- Quality fluctuations
- Multiple machine units for quasicontinuous needed
- High maintenance



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Residence time: < 10 min Maximum solids content: 90 wt.-% ^[1]

- Energy efficient due to short resicence time
- Less investment costs
- Continuous dispersing and coating
- Higher solids contents possible
- Water based processes are more feasible



VS.

Sustainable electrode production iPA Direct tape extrusion and granule based coating Institute for Particle Technology ₩ Direct tape **Coating Unit** extrusion (high \equiv viscous pastes) 0 **Twin-Screw Extruder** 0 IR- drying zone Single-Screw-Extruder releasefoil unit Wet granules 0 0 granulate IR- drying zone dosage based coating methode (0) 0 Twin screw extruder calender Technische 26.08.2021 | Arno Kwade | Circular Battery Cell Production | Slide 16 Universität BATTERY Braunschweig LABFACTORY

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High viscous manufacturing of elektrodes

Potentials and novelty

- Less investment and drying costs
- Less energy requirement and better GWP
- Time and location decoupling of dispersion and coating possible
- Increased initial electrode density
- No migration of binder and conductivity additives





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Process based on wet granules First test cell results





Standard process route:

- Pore systems are formed by the evaporation of the solvent
- Assumption: Inhomogeneous compression of pore channels due to calendering
- → The diffusive resistance in certain areas of the electrode is increased

Granule based process route:

- More homogenous pore and material structure
- Higher initial capacity of high viscous processed electrode





Sustainable electrode production Solvent-free dry coating as an alternative Dry multistep mixing Fine dosing Hot pressing treatment Cell production, electrochemically analysis Finished electrode

- Reducing required heating energy up to 90 %
- Saving space due to shorter thermal sections, pointless waste gas treatment
- Reducing dwell time due to faster thermal process





Potentials of sustainable battery cell manufacturing

Process chain of lithium-ion battery cell manufacturing



battery production technologies. Nature Energy, vol. 3, pages 290-300 (2018)

Investigation of different post-drying intensities



Post-drying (PD) methods:

- Argon-PD: 3x vacuum + 3x Argon purging in lock of glovebox (15min)
- Medium-vac.-PD: 120°C / 18 h in vacuum, followed by manual purging (3 argon/vac. cycles)
- Long-vac.-PD : 120°C / 96 h in vacuum, followed by manual purging (3 argon/vac. cycles)
- → Argon-PD already considerably reduces the remaining water content of total cells





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[1] F. Huttner, W. Haselrieder, A. Kwade, The Influence of Different Post-Drying Procedures on Remaining Water Content and Physical and Electrochemical Properties of Lithium-Ion Batteries. *Energy Technology* (2019). doi/10.1002/ente.201900245.

Rising intensity



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Investigation of different post-drying intensities for LIBs

- Mild Argon-PD shows best cell performance despite highest moisture
- Cell performance not only dependent on residual moisture, but also significantly on post-drying intensity
- → Post-drying parameters must be chosen gentle enough to maintain sensitive binder network of electrodes
- Argon-PD: 3x vacuum / Argon purging (15min)
- Medium-vac.-PD: electr.: 120°C / 18 h in vac., sep.: 60°C / 4 h
- Long-vac.-PD : electr.: 120°C / 96 h in vac., sep.: 60°C / 4 h





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Sustainable cell production and conditioning Material and energy flows in cell manufacturing Institute for Particle Technology ~ 45% to 60% ~ 35% to 55% Electricity Anode Dry Cu – current Room Electrolyte collector NMC Zel Zellherstellung **Benchbatt** NMP (Solvent) Cathode GEFÖRDERT VOM Bundesministeriun für Bildung und Forschung Mini-Environments instead of large dry rooms Technische 26.08.2021 | Arno Kwade | Circular Battery Cell Production | Slide 23 Universität BATTERY Braunschweig Thomitzek M, Cerdas F, Thiede S, Herrmann C (2019) Cradle-to-Gate Analysis of the LABFACTORY

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Embodied Energy in Lithium Ion Batteries. Procedia CIRP. 80:304-309.

Sustainable cell production

Impact of accuracy in cell assembly



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Cost breakdown production and material costs C//NMC 111, PHEV2, 36 Ah





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J.-H. Schünemann – Modell zur Bewertung der Herstellkosten von Lithiumionenbatteriezellen, Dissertation, iPAT Schriftenreihe, Sierke-Verlag

Cumulative cost over production steps C//NMC, PHEV2, 36 Ah

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J.-H. Schünemann – Modell zur Bewertung der Herstellkosten von Lithiumionenbatteriezellen, Dissertation, iPAT Schriftenreihe, Sierke-Verlag





Requirements in process chain



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Competence Cluster Recycling & Green Battery Action – Recycling of electrode production rejects





Reuse and recycling of production scrap Electrode scrap



automotive battery production technologies. Nature Energy, vol. 3, pages 290-300 (2018)

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Reuse and recycling of production scrap

Recycling of electrode scrap

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Dry-mechanical





Reuse and recycling of production scrap Cell scrap Institute for Particle Technology **Electrochemical** production Electrode-**Quality Control** not functional **Pretreatment** Coating and Material Calendering and Mixing Drying inadequate performance production Cell-Enclosure Cutting and Reuse Packaging Contacting and Drying Welding Processing to raw materials conditioning alternative Cellapplications, e.g. Formation Electrochemical Filling Cells stationary energy **Quality Control** and Ageing storage Technische 26.08.2021 | Arno Kwade | Circular Battery Cell Production | Slide 33 Universität BATTERY Braunschweig LABFACTORY According to: Kwade, A. and Haselrieder, W. et al.: Current status and challenges for + BRAUNSCHWEIG automotive battery production technologies. Nature Energy, vol. 3, pages 290-300 (2018)

Circulating production scrap and EoL-Batteries Recycling Technologies





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Competence Cluster Recycling & Green Battery





Mission:

Designing an energy and material efficient battery life cycle by closing material loops

Cluster goals:

Development and application of innovative recycling and resynthesis processes

- Increasing the quality and availability of data for the development of multidisciplinary life cycle models and tools
- Recommendations for design for recycling and for end-ofuse

Funding:

approx. 30 million €









Battery Recycling and Circular Battery Production



- Cycle materials and keep them in country – less CO₂, negative social impact and costs
- Be competitive in battery cell production by utilizing the
 75% material cost portion
- Enable a sustainable future mobility

Technische Universität Braunschweig





Battery LabFactory Braunschweig Circular Battery Economy at BLB







Pilot scale circular battery production facility









CEID Working Group Traction Batteries: Circular Economy Focus and Institute for Particle Technology Tr Production of Module and Raw material Cell **Collection and** Recovery/ Refinement active materials system Application Disposal extraction production Recycling return and precursors production Repair/ Other value Refurbishment intensifying chain Recycling Recycling use (i.e. networks V2G, car-Re-use sharing) (Second Life) 6 Recycling Higher value re-use in other applications www.circular-economy-initiative.de Source: Circular Economy Initiative Deutschland; based on World Economic Forum 2019 Technische 26.08.2021 | Arno Kwade | Circular Battery Cell Production | Slide 38 Universität BATTERY Braunschweig LABFACTORY

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Pilot Profile I: "Knowledge of battery life"

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Information flows to promote the recycling of lithium-ion batteries

 Source: Circular Economy Initiative Deutschland; based on World Economic Forum 2019
 www.circular-economy-initiative.de

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Pilot Profile III: "Disassembly network"





Source: Circular Economy Initiative Deutschland



Recommendation of recovery rates



Circular Economy Initiative Deutschland EU directive (proposal)				
Material	Recommended Recovery rates*			
	2025 – binding	2030 – to be aspired to ^{•••,•••••}	2025	2030
Total battery**	60 %	70 %	65 %	70%
Lithium	50 %	85 %	35 %	70 %
Cobalt	85 %	90 %	90 %	95 %
Nickel	85 %	90 %	90 %	95 %
Copper	85 %	90 %	90 %	95 %
Steel	90 %	95 %	-	-
Aluminum (without Al foil)	90 %	95 %	-	-



Source: Circular Economy Initiative Deutschland; based on World Economic Forum 2019

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www.circular-economy-initiative.de

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Lithium-ion based all-solid-state-battery - basics



- Liquid electrolyte and porous separator are replaced by a solid electrolyte
- Li-metal or Li-free anode
- Main theoretical/potential advantages:
 - Broader temperature window
 - Higher energy density potential (Wh/L)
 - Potential of higher ionic conductivity for fast-charging capability
 - Increased safety
- Solid electrolyte
- Active material
- Conductive carbon
- Binder
- Lithium ions







Performance potential of all solid state batteries



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 - Potential of higher ionic conductivity for fast-charging capability
 - Increased safety



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[1] S. Randau, D. A. Weber, O. Kötz, R. Koerver, P. Braun, A. Weber, E. Iver-Tiffée, T. Adermann, J. Kulisch, W. Zeier, F. H. Richter, and J. Janek, "Benchmarking the performance of all-solid-state lithium batteries" Nat. Energy 5, S. 259-270. 2020.





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Challenges of high-energy ASSB

Li-metal anode

- Thin (10 µm) for opt. energy density
- High surface area for good discharge rate
- Compensation for volume expansion
- Prevention of Li-dendrite formation





Separator

- Electrochemically stable against Li-metal
- High ionic conductivity
- Mechanically stable against Lidendrites
- Thin enough for high energy density

Cathode challenges:

- Volume expansion of active material
- Electrical and ionic percolation
- Interface resistances (ionic and electric phases)





Production of solid state batteries

Challenges:

- Individual processing routes required for different solid electrolytes
 - Polymers: mechanical and thermal degradation of polymers limits processing windows for dispersion and extrusion
 - Sulfides: processing sensitive to water/ various solvents
 - Oxides: thermal processing window exceeds degradation threshold of active materials, conductivity decreased with decreasing sintering temperature
- Lithium-anode production is demanding and requires novel processing techniques







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Solvent free production chains of polymer based composite cathodes for solid state batteries





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Electrochemical Performance



- iPAT S
- Polymer electrolyte distribution shows strong influence on ion. conductivity
- Input PEO mol. weight effects viscosity during processing and therefore the carbon black distribution







LH1

Benchmarking the Perfomance





Reach Target region by:

- Increase area capacity and use multy-layer electrodes to reduce transport limitations
- Use hybrid electrolytes with low amount of oxides or sulfides to lower operating temperature and transport limitations
- Enable the use of NCM as cathode active material



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Recycling target:

Economically and ecologically efficient recovery of high quality products or precursors for battery production

Challenges:

- High number of materials used and lower material value reduces ecological incentive and increase the recycling effort
- High safety requirements for the processes due to reactivity of various components (Lithium, sulfide based SE)







Anode:

Recovery of lithium or lithium salts

Challenges and opportunities:

- Reactivity of lithium with water and air
- ✓ Higher concentration of lithium in SSB
- ✓ Possible solution step to recover lithium salts
- ✓ In aqueous systems, formation of LiOH with release of H2
- ✓ lithum-free anodes lead to simplified recycling due to absence of metallic lithium











Solid Electrolyte (SE):

Direct Recovery of SE or Recovery of precursors

Challenges and opportunities:

- Different cell chemistries require different recycling methods
- Reactivity of sulfide based solid electrolytes
- Hybrid SSBs increase the recycling effort
- ✓ Direct Recycling methods due to the absence of chemical degradation for oxide based SE
- ✓ Solution Processes for sulfide and polymer based SE









Composite cathode:

Recovery of active material and SE

Challenges and opportunities:

- Selective Separation necessary
- Solid-solid structure makes separation into pure fractions difficult
- Polymer SSB: low material values reduce ecological incentive
- ✓ Hydrometallurgical approaches to recover both, SE and active materials





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Conclusion

- Sustainable production processes and circluar battery production with closed material cycles are very important for long term success of electro mobility
- Environmental impact of battery systems (e.g. GWP in t CO₂/kWh) must be decreased to reduce distance before EV have a lower GWP than conventional cars
- Internal recycling of production scrap is a very important measure to improve CO2-footpring of cell production beside usage of energy efficient processes and green energy
- Battery materials should be chosen not only regarding their performance, but also regarding the environmental impact
- Solid state batteries have the potential to increase the cell performance further, but will increase the effort for recycling of production scrap and EoL batteries



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Thank you very much

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....for the great work of my PhD-students and co-workers

... for your attention





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